

# Periodic Table of the Elements

	57	La	Cerium 140.116	58	Ce	Praseodymium 140.908	59	Pr	Neodymium 144.243	60	Nd	Promethium	144.913	61	Pm		62	Sm	Samarium 150.36	63	Gd	Erbium 167.259	64	Ho	Holmium 168.934	65	Tb	Dysprosium 151.964	66	Dy	Terbium 158.925	67	Er	Gadolinium 162.500	68	Tm	Thulium 157.25	69	Yb	Erbium 164.930	70	Lu	Ytterbium 173.055	71		Lutetium 174.967
	89	Ac	Actinium 227.028	90	Th	Thorium 232.038	91	Pa	Protactinium 231.036	92	U	Uranium 238.029	93	Np	Neptunium 237.048	94	Pu	Plutonium	95	Cf	Americium 243.061	96	Bk	Berkelium 247.070	97	Am	Curium 251.080	98	Fm	Einsteinium 247.095	99	Md	Mendelevium [254]	100	Es	Californium 258.1	101	Cm	Nobelium 259.101	102	No	Lawrencium [262]	103	Lr		



## ★ आवर्त सारणी (Periodic Table) के प्रमुख वैज्ञानिक एवं उनके वर्ष ★

### 1. डोबेराइनर (Johann Wolfgang Döbereiner) – 1817

👉 "त्रिक का सिद्धांत (Law of Triads)" प्रस्तुत किया।

### 2. न्यूलैंड्स (John Newlands) – 1864

👉 "अष्टक नियम (Law of Octaves)" दिया।

### 3. लोथर मेयर (Lothar Meyer) – 1869

👉 तत्वों के परमाणु आयतन और परमाणु भार के संबंध को समझाया।

### 4. मेंडलीफ (Dmitri Ivanovich Mendeleev) – 1869

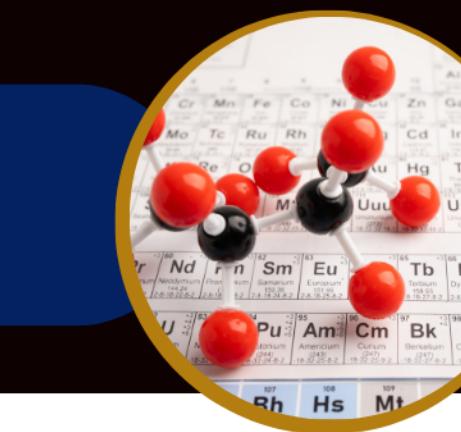
👉 पहली संपूर्ण आवर्त सारणी (Complete Periodic Table) दी।

### 5. हॉसले (Henry Moseley) – 1913

👉 "परमाणु क्रमांक (Atomic Number)" के आधार पर आवर्तन नियम दिया।



Cr Chromium 2.00	Mn Manganese 54.94	Fe Iron 55.85	Co Cobalt 59	Ni Nickel 69	Cu Copper 63.55	Zn Zinc 65.38	Ga Gallium 69.72	Ge Germanium 72.63	Sr Strontium 82.05, 82.09	Cl Chlorine 35.44, 35.46	Ar Argon 39.95
42 Mo Molybdenum 96	43 Tc Technetium (98)	44 Ru Ruthenium 101	47 Ag Silver 107.9	48 Cd Cadmium 112.4	49 In Indium 114.8	50 Sn Tin 118.7	51 Sb Antimony 121.8	52 Te Tellurium 127.6	53 I Iodine 126.9	36 Kr Krypton 83.80	
74 W Tungsten 183.8	75 Re Rhenium 186.2	79 J Joliotium (107)	80 Hg Mercury 200.6	81 Tl Thallium 204.3, 204.4	82 Pb Lead 207.2	83 Bi Bismuth 209.0	84 Po Polonium (209)	85 At Astatine (210)	86 Rn Radon (222)	118 Og Oganesson (294)	
96 Bh Bohrium (270)	107 Bh Bohrium (270)	112 Cn Copernicium (285)	113 Nh Nihonium (284)	114 Fl Flerovium (289)	115 Mc Moscovium (288)	116 Lv Livermorium (293)	117 Ts Tennessine (294)	118 Og Oganesson (294)	70 Yb Ytterbium 173.1	71 Lu Lutetium 175.0	
66 Dy Dysprosium 162.5	67 Ho Holmium 164.9	68 Er Erbium 167.3	69 Tm Thulium 168.9	98 Fr Francium )	99 Es Einsteinium (252)	100 Fm Fermium (257)	101 Md Mendelevium (258)	102 No Nobelium (259)	103 Lr Lawrencium (262)		



👉 सन् 1913 में वैज्ञानिक **मॉसले (Moseley)** ने तत्वों का अध्ययन किया।

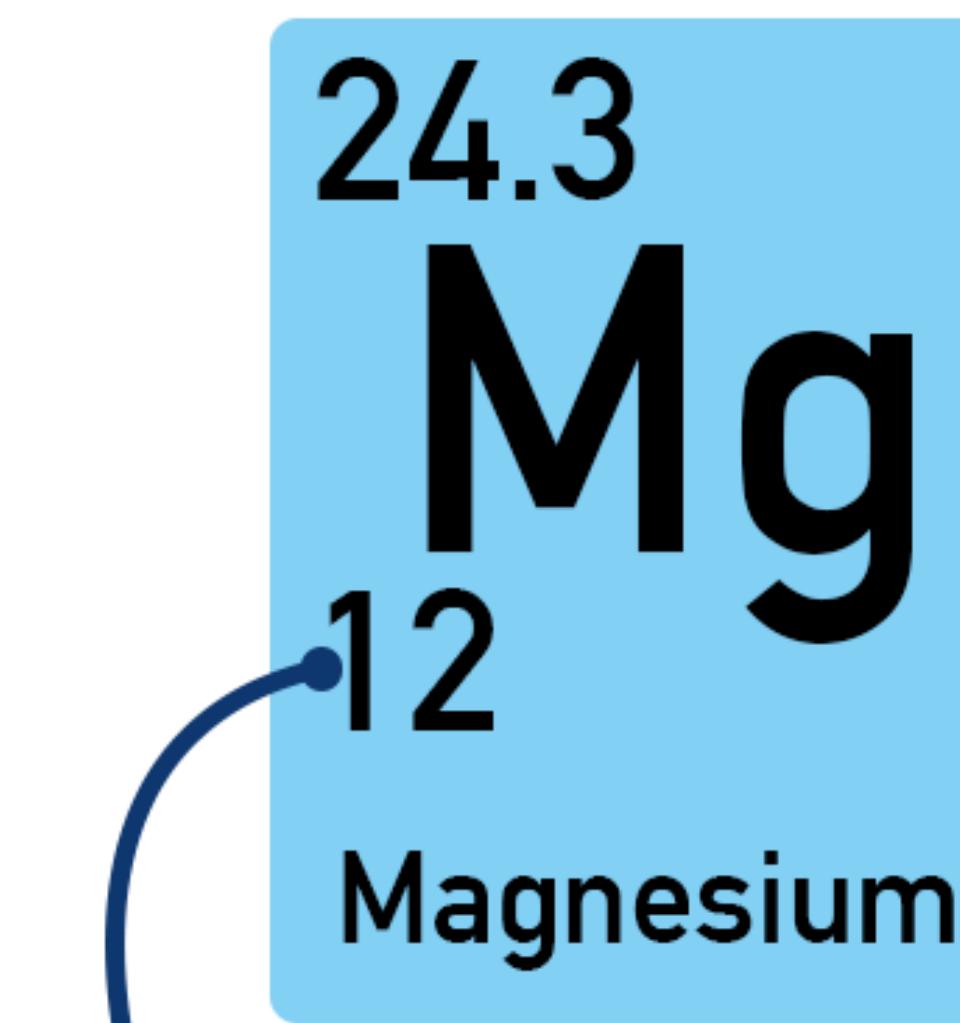
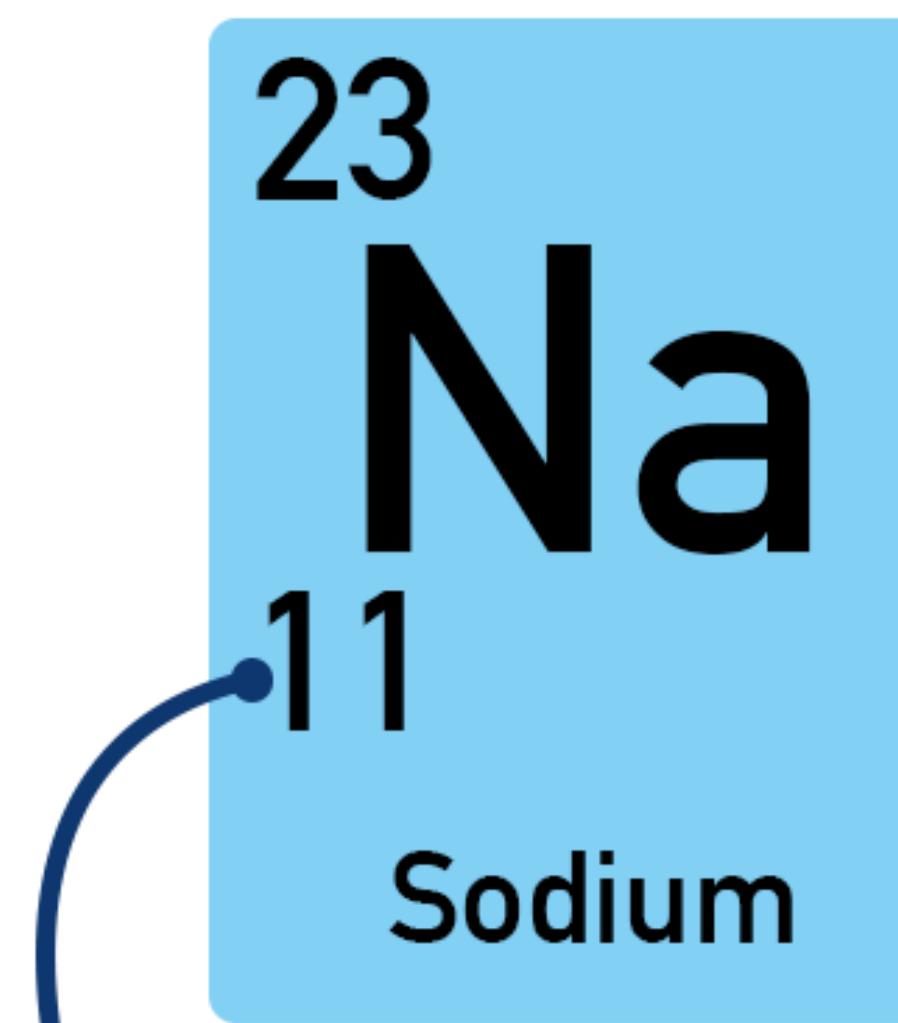
→ In 1913, scientist **Henry Moseley** studied the elements.

👉 उन्होंने एक नया गुण खोजा जिसे **परमाणु संख्या (Atomic Number)** कहा गया।

→ He discovered a new property called the **Atomic Number**.

👉 इस खोज के आधार पर उन्होंने **आधुनिक आवर्त नियम (Modern Periodic Law)** दिया।

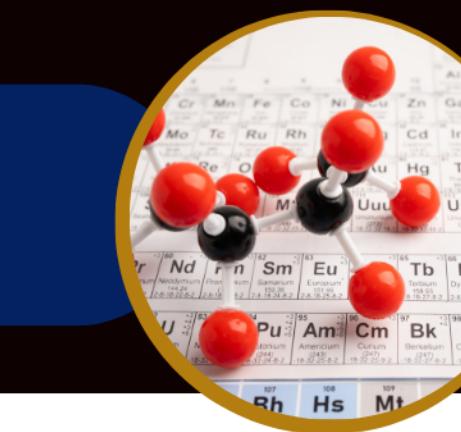
→ Based on this discovery, he gave the **Modern Periodic Law**.



Atomic number  
= 11 protons

Atomic number  
= 12 protons





## ■ आधुनिक आवर्त नियम का कथन / Statement Of Modern Periodic Law:

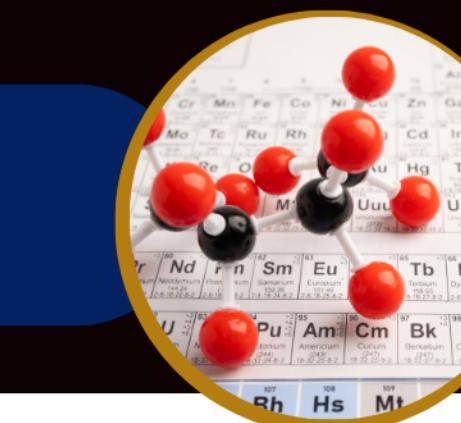
“👉 तत्वों के भौतिक और रासायनिक गुण उनकी परमाणु संख्याओं के अनुसार आवर्तित होते हैं।”

→👉 “The Physical And Chemical Properties Of Elements Are Periodic Functions Of Their Atomic Numbers.”



H																									He
1 (1.01)																								2 (4.00)	
Li	Be																								
3 (6.94)	4 (9.01)																								
Na	Mg																								
11 (22.99)	12 (24.30)																								
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr								
19 (39.10)	20 (40.08)	21 (44.95)	22 (47.88)	23 (50.94)	24 (51.99)	25 (54.94)	26 (55.85)	27 (58.93)	28 (58.69)	29 (63.55)	30 (65.36)	31 (69.72)	32 (72.59)	33 (74.92)	34 (78.96)	35 (79.90)	36 (83.80)								
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe								
37 (85.47)	38 (87.62)	39 (88.90)	40 (91.22)	41 (92.91)	42 (95.94)	43 (98.91)	44 (101.07)	45 (102.90)	46 (107.87)	47 (107.87)	48 (112.41)	49 (114.82)	50 (118.69)	51 (121.75)	52 (127.60)	53 (126.90)	54 (131.29)								
Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn								
55 (132.90)	56 (137.33)	57 (138.90)	72 (178.49)	73 (180.95)	74 (183.85)	75 (186.21)	76 (190.20)	77 (192.22)	78 (195.08)	79 (196.97)	80 (200.59)	81 (204.38)	82 (207.20)	83 (208.98)	84 (209.99)	85 (220.02)	86 (222.02)								
Fr	Ra	Ac	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn														
87 (223.02)	88 (226.02)	89 (227.03)	104 (281.10)	105 (262.11)	106 (263.12)	107 (264.12)	108 (277.18)	109 (268.14)	110 (268.14)	111 (280*)	112 (285*)														
												Fl													
												114 (289*)													
												116 (297*)													

Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
58 (140.11)	59 (140.91)	60 (144.24)	61 (146.92)	62 (150.38)	63 (151.96)	64 (157.25)	65 (158.93)	66 (162.50)	67 (164.93)	68 (167.26)	69 (166.93)	70 (173.04)	71 (174.97)
Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr
90 (232.04)	91 (231.04)	92 (238.05)	93 (237.05)	94 (244.06)	95 (243.06)	96 (247.07)	97 (247.07)	98 (251.08)	99 (252.08)	100 (257.10)	101 (259.10)	102 (259.10)	103 (262.11)



# परमाणु संख्या ही क्यों Why Only Atomic Number

 किसी तत्व की परमाणु संख्या उस तत्व के एक परमाणु में

**उपस्थित प्रोटॉनों अथवा इलेक्ट्रॉनों की संख्या के बराबर होती है**

→ The **Atomic Number** Of An Element Is Equal To The Number  
Of Protons Or Electrons Present In One Atom Of That Element

 किसी तत्व के लिए इसका मान हमेशा स्थिर रहता है

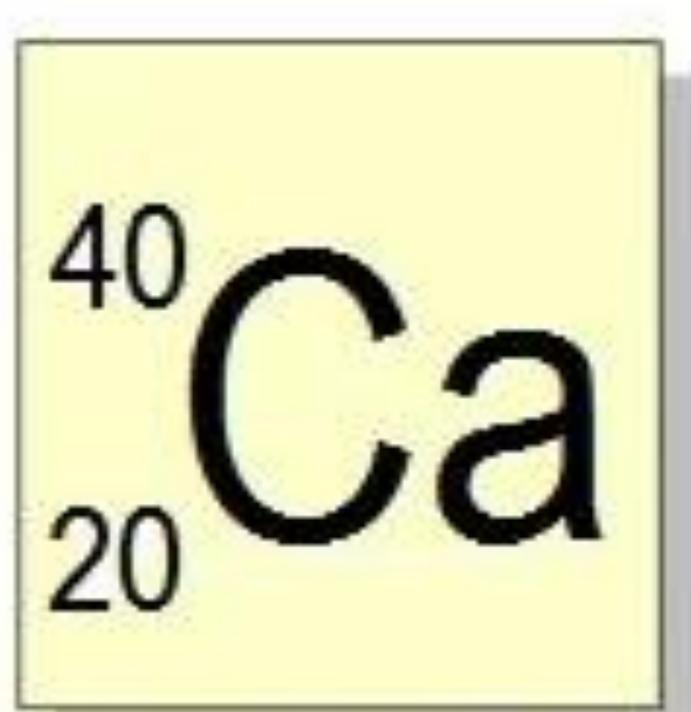
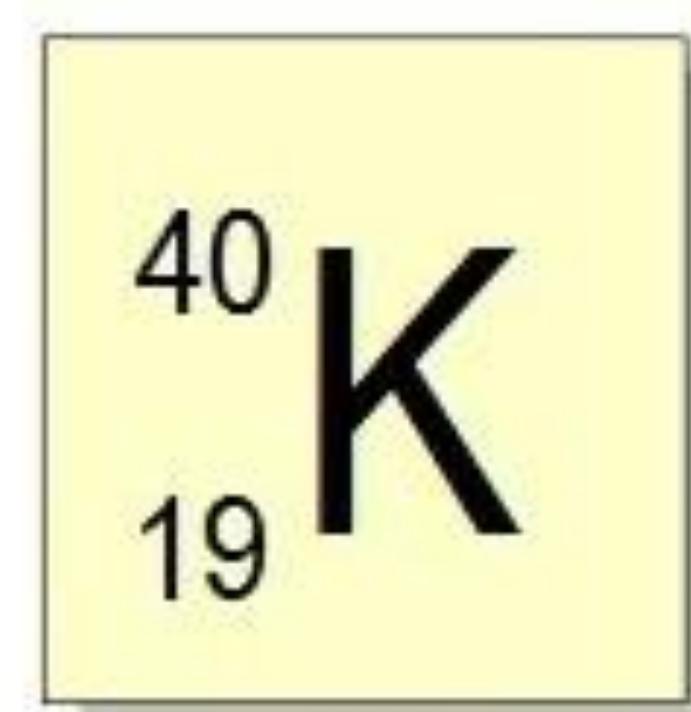
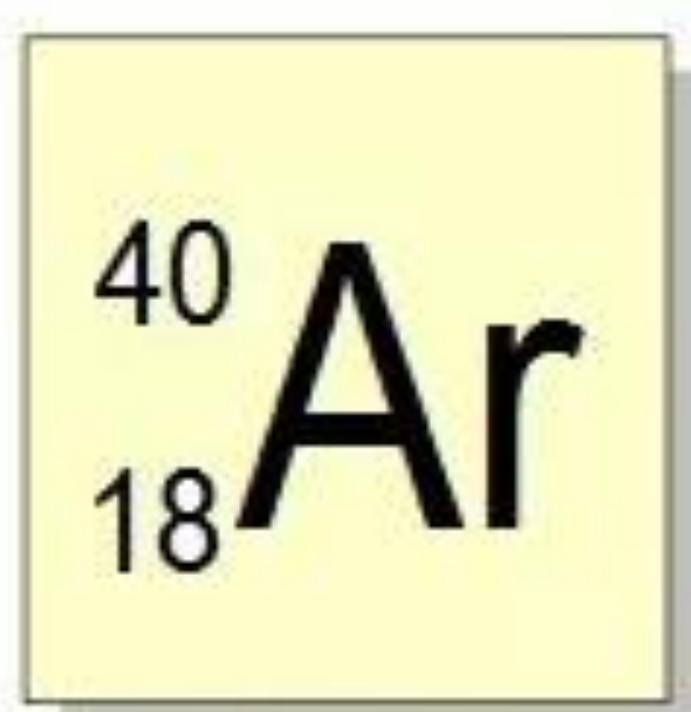
→ Its Value Is **Always Constant** For Any Element

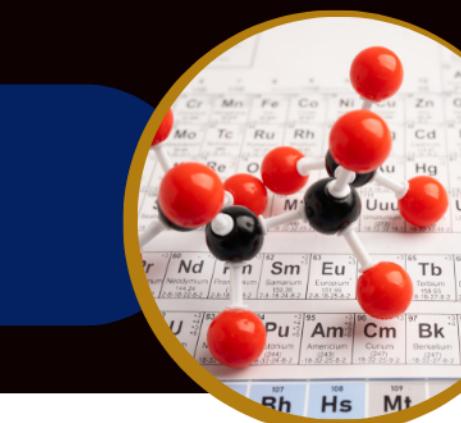
## दो अलग-अलग तत्वों की परमाणु संख्या कभी समान नहीं होती

→ The Atomic Number Of Two Different Elements Is Never The Same.

H 1 (1.01)																	He 2 (4.00)
Li 3 (6.94)	Be 4 (9.01)																
Na 11 (22.99)	Mg 12 (24.30)																
K 19 (39.10)	Ca 20 (40.08)	Sc 21 (44.95)	Ti 22 (47.88)	V 23 (50.94)	Cr 24 (51.99)	Mn 25 (54.94)	Fe 26 (55.85)	Co 27 (58.93)	Ni 28 (58.69)	Cu 29 (63.55)	Zn 30 (65.38)	Ga 31 (69.72)	Ge 32 (72.59)	As 33 (74.92)	Se 34 (78.96)	Br 35 (79.90)	Kr 36 (83.80)
Rb 37 (85.47)	Sr 38 (87.62)	Y 39 (88.90)	Zr 40 (91.22)	Nb 41 (92.91)	Mo 42 (95.94)	Tc 43 (98.91)	Ru 44 (101.07)	Rh 45 (102.90)	Pd 46 (107.87)	Ag 47 (107.87)	Cd 48 (112.41)	In 49 (114.82)	Sn 50 (118.69)	Sb 51 (121.75)	Te 52 (127.60)	I 53 (126.90)	Xe 54 (131.29)
Cs 55 (132.90)	Ba 56 (137.33)	La 57 (138.90)	Hf 72 (178.49)	Ta 73 (180.95)	W 74 (183.85)	Re 75 (186.21)	Os 76 (190.20)	Ir 77 (192.22)	Pt 78 (196.08)	Au 79 (196.97)	Hg 80 (200.59)	Tl 81 (204.38)	Pb 82 (207.20)	Bi 83 (208.98)	Po 84 (208.98)	At 85 (209.99)	Rn 86 (222.02)
Fr 87 (223.02)	Ra 88 (226.02)	Ac 89 (227.03)	Rf 104 (261.10)	Db 105 (262.11)	Sg 106 (263.12)	Bh 107 (264.12)	Hs 108 (277.13)	Mt 109 (268.14)	Ds 110 (268.14)	Rg 111 (280*)	Cn 112 (285*)		Fl 114 (289*)		Lv 116 (297*)		

Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
58 (140.11)	59 (140.91)	60 (144.24)	61 (146.92)	62 (150.36)	63 (151.96)	64 (157.25)	65 (158.93)	66 (162.50)	67 (164.93)	68 (167.26)	69 (168.93)	70 (173.04)	71 (174.97)
Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr
90 (232.04)	91 (231.04)	92 (238.05)	93 (237.05)	94 (244.06)	95 (243.06)	96 (247.07)	97 (247.07)	98 (251.08)	99 (252.08)	100 (257.10)	101 (258.10)	102 (259.10)	103 (262.11)





## आधुनिक आवर्त सारणी The modern periodic table

आवर्त सारणी में कुल 7 क्षैतिज कतारे (आवर्त / Periods) होती हैं।

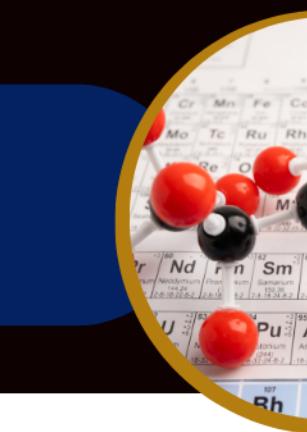
→ There Are 7 Horizontal Rows (Called Periods) In The Periodic Table.

इसमें कुल 18 ऊर्ध्व स्तम्भ (वर्ग / Groups) होते हैं।

→ It Has 18 Vertical Columns (Called Groups).

H 1 (1.01)																		He 2 (4.00)
Li 3 (6.94)	Be 4 (9.01)																	
Na 11 (22.99)	Mg 12 (24.30)																	
K 19 (39.10)	Ca 20 (40.08)	Sc 21 (44.95)	Ti 22 (47.88)	V 23 (50.94)	Cr 24 (51.99)	Mn 25 (54.94)	Fe 26 (55.85)	Co 27 (58.93)	Ni 28 (58.69)	Cu 29 (63.55)	Zn 30 (65.38)	Ga 31 (69.72)	Ge 32 (72.58)	As 33 (74.92)	Se 34 (78.98)	Br 35 (79.90)	Kr 36 (83.80)	
Rb 37 (85.47)	Sr 38 (87.63)	Y 39 (88.90)	Zr 40 (91.22)	Nb 41 (92.91)	Mo 42 (95.94)	Tc 43 (98.91)	Ru 44 (101.07)	Rh 45 (102.90)	Pd 46 (107.87)	Ag 47 (107.87)	Cd 48 (112.41)	In 49 (114.82)	Sn 50 (118.69)	Sb 51 (121.75)	Te 52 (127.60)	I 53 (126.90)	Xe 54 (131.29)	
Cs 55 (132.90)	Ba 56 (137.33)	La 57 (138.90)	Hf 72 (178.49)	Ta 73 (180.95)	W 74 (183.85)	Re 75 (186.21)	Os 76 (190.20)	Ir 77 (192.22)	Pt 78 (196.08)	Au 79 (196.97)	Hg 80 (200.59)	Tl 81 (204.38)	Pb 82 (207.20)	Bi 83 (208.98)	Po 84 (209.99)	At 85 (222.02)	Rn 86 (222.02)	
Fr 87 (223.02)	Ra 88 (226.02)	Ac 89 (227.03)	Rf 104 (261.10)	Db 105 (262.11)	Sg 106 (263.12)	Bh 107 (264.12)	Hs 108 (277.13)	Mt 109 (288.14)	Ds 110 (288.14)	Rg 111 (280*)	Cn 112 (280*)	Fl 114 (280*)		Lv 116 (297*)				

Ce 58 (140.11)	Pr 59 (140.91)	Nd 60 (144.24)	Pm 61 (146.92)	Sm 62 (150.36)	Eu 63 (151.96)	Gd 64 (157.25)	Tb 65 (158.93)	Dy 66 (162.50)	Ho 67 (164.83)	Er 68 (167.26)	Tm 69 (168.93)	Yb 70 (173.04)	Lu 71 (174.97)
Th 90 (232.04)	Pa 91 (231.04)	U 92 (238.05)	Np 93 (237.05)	Pu 94 (244.08)	Am 95 (243.06)	Cm 96 (247.07)	Bk 97 (247.07)	Cf 98 (254.08)	Es 99 (252.08)	Fm 100 (257.10)	Md 101 (259.10)	No 102 (262.11)	Lr 103 (262.11)



H 1 (1.01)															He 2 (4.00)		
Li 3 (6.94)	Be 4 (9.01)																
Na 11 (22.99)	Mg 12 (24.30)																
K 19 (39.10)	Ca 20 (40.08)	Sc 21 (44.95)	Ti 22 (47.88)	V 23 (50.94)	Cr 24 (51.98)	Mn 25 (54.94)	Fe 26 (55.85)	Co 27 (58.93)	Ni 28 (58.69)	Cu 29 (63.55)	Zn 30 (65.38)	Ga 31 (69.72)	Ge 32 (72.58)	As 33 (74.82)	Se 34 (78.96)	Br 35 (79.90)	Kr 36 (83.80)
Rb 37 (85.47)	Sr 38 (87.62)	Y 39 (88.90)	Zr 40 (91.22)	Nb 41 (92.91)	Mo 42 (95.94)	Tc 43 (98.91)	Ru 44 (101.07)	Rh 45 (102.90)	Pd 46 (107.87)	Ag 47 (107.87)	Cd 48 (112.41)	In 49 (114.82)	Sn 50 (118.69)	Sb 51 (121.75)	Te 52 (127.60)	I 53 (126.90)	Xe 54 (131.29)
Cs 55 (132.90)	Ba 56 (137.33)	La 57 (138.90)	Hf 72 (178.49)	Ta 73 (180.95)	W 74 (183.85)	Re 75 (186.21)	Os 76 (190.20)	Ir 77 (192.22)	Pt 78 (195.08)	Au 79 (196.97)	Hg 80 (200.59)	Tl 81 (204.38)	Pb 82 (207.20)	Bi 83 (208.98)	Po 84 (209.99)	At 85 (222.02)	Rn 86 (222.02)
Fr 87 (223.02)	Ra 88 (226.02)	Ac 89 (227.03)	Rf 104 (261.10)	Db 105 (262.11)	Sg 106 (263.12)	Bh 107 (264.12)	Hs 108 (277.13)	Mt 109 (268.14)	Ds 110 (268.14)	Rg 111 (280*)	Cn 112 (285*)		Fl 114 (289*)		Lv 116 (297*)		

Ce 58 (140.11)	Pr 59 (140.91)	Nd 60 (144.24)	Pm 61 (146.92)	Sm 62 (150.36)	Eu 63 (151.96)	Gd 64 (157.25)	Tb 65 (158.93)	Dy 66 (162.50)	Ho 67 (164.93)	Er 68 (167.26)	Tm 69 (168.93)	Yb 70 (173.04)	Lu 71 (171.97)
Th 90 (232.04)	Pa 91 (231.04)	U 92 (238.05)	Np 93 (237.05)	Pu 94 (244.06)	Am 95 (243.06)	Cm 96 (247.07)	Bk 97 (247.07)	Cf 98 (251.08)	Es 99 (252.08)	Fm 100 (257.10)	Md 101 (258.10)	No 102 (259.10)	Lr 103 (262.11)



आवर्त	तत्वों की संख्या
1	2
2	8
3	8
4	18
5	18
6	32
7	शेष तत्व

H 1 (1.01)																			He 2 (4.00)
Li 3 (6.94)	Be 4 (9.01)																		
Na 11 (22.99)	Mg 12 (24.30)																		
K 19 (39.10)	Ca 20 (40.08)	Sc 21 (44.95)	Ti 22 (47.88)	V 23 (50.94)	Cr 24 (51.99)	Mn 25 (54.94)	Fe 26 (55.85)	Co 27 (58.93)	Ni 28 (58.69)	Cu 29 (63.55)	Zn 30 (65.38)	Ga 31 (69.72)	Ge 32 (72.59)	As 33 (74.92)	Se 34 (78.96)	Br 35 (79.90)	Kr 36 (83.80)		
Rb 37 (85.47)	Sr 38 (87.62)	Y 39 (88.90)	Zr 40 (91.22)	Nb 41 (92.91)	Mo 42 (95.94)	Tc 43 (98.91)	Ru 44 (101.07)	Rh 45 (102.90)	Pd 46 (107.87)	Ag 47 (107.87)	Cd 48 (112.41)	In 49 (114.82)	Sn 50 (118.69)	Sb 51 (121.76)	Te 52 (126.90)	I 53 (126.90)	Xe 54 (131.29)		
Cs 55 (132.90)	Ba 56 (137.33)	La 57 (138.90)	Hf 72 (178.49)	Ta 73 (180.95)	W 74 (183.85)	Re 75 (186.21)	Os 76 (190.20)	Ir 77 (192.22)	Pt 78 (195.08)	Au 79 (196.97)	Hg 80 (200.59)	Tl 81 (204.38)	Pb 82 (207.20)	Bi 83 (208.98)	Po 84 (209.99)	At 85 (209.99)	Rn 86 (222.02)		
Fr 87 (223.02)	Ra 88 (226.02)	Ac 89 (227.03)	Rf 104 (261.10)	Db 105 (262.11)	Sg 106 (263.12)	Bh 107 (264.12)	Hs 108 (277.13)	Mt 109 (268.14)	Ds 110 (268.14)	Rg 111 (280*)	Cn 112 (285*)		Fl 114 (289*)		Lv 116 (297*)				

Ce 58 (140.11)	Pr 59 (140.91)	Nd 60 (144.24)	Pm 61 (146.92)	Sm 62 (150.38)	Eu 63 (151.96)	Gd 64 (157.25)	Tb 65 (158.93)	Dy 66 (182.50)	Ho 67 (164.93)	Er 68 (167.26)	Tm 69 (168.93)	Yb 70 (173.04)	Lu 71 (174.97)					
Th 90 (232.04)	Pa 91 (231.04)	U 92 (238.05)	Np 93 (237.05)	Pu 94 (244.05)	Am 95 (243.06)	Cm 96 (247.07)	Bk 97 (247.07)	Cf 98 (251.08)	Es 99 (252.08)	Fm 100 (257.10)	Md 101 (258.10)	No 102 (259.10)	Lr 103 (262.11)					



• (1) S-ब्लॉक तत्व (S-block Elements)

H 1 (1.01)		
Li 3 (6.94)	Be 4 (9.01)	
Na 11 (22.99)	Mg 12 (24.30)	

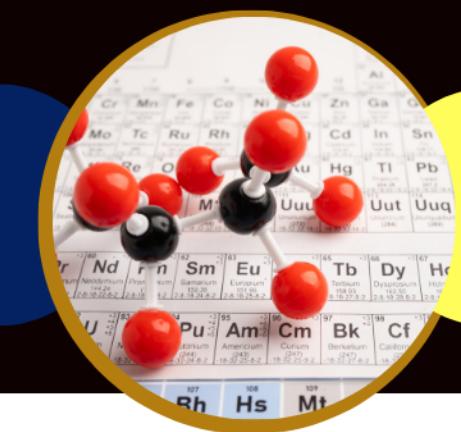
He 2 (4.00)						
B 5 (10.81)	C 6 (12.01)	N 7 (14.01)	O 8 (15.99)	F 9 (18.99)		
Al 13 (26.98)	Si 14 (28.09)	P 15 (30.97)	S 16 (32.06)	Cl 17 (35.45)	Ar 18 (39.95)	
Ga 31 (69.72)	Ge 32 (72.59)	As 33 (74.92)	Se 34 (78.96)	Br 35 (79.90)	Kr 36 (83.80)	
In 31 (114.82)	Sn 50 (118.89)	Sb 51 (121.75)	Te 52 (121.60)	I 53 (126.90)	Xe 54 (131.29)	
Tl 81 (204.38)	Pb 82 (207.20)	Bi 83 (208.98)	Po 84 (209.99)	At 85 (222.02)	Rn 86 (222.02)	
Fl 114 (289*)	Lv 116 (297*)					

• (2) P-ब्लॉक तत्व (P-block Elements)

K 19 (39.10)	Ca 20 (40.08)	Sc 21 (44.95)	Ti 22 (47.86)	V 23 (50.94)	Cr 24 (51.99)	Mn 25 (54.94)	Fe 26 (55.86)	Co 27 (56.93)	Ni 28 (58.69)	Cu 29 (63.55)	Zn 30 (65.38)	Ga 31 (69.72)	Ge 32 (72.59)	As 33 (74.92)	Se 34 (78.96)	Br 35 (79.90)	Kr 36 (83.80)
Rb 37 (85.47)	Sr 38 (87.62)	Y 39 (88.90)	Zr 40 (89.22)	Nb 41 (92.91)	Mo 42 (95.94)	Tc 43 (98.91)	Ru 44 (101.97)	Rh 45 (102.90)	Pd 46 (107.87)	Ag 47 (107.87)	Cd 48 (112.41)	In 49 (114.82)	Sn 50 (118.89)	Sb 51 (121.75)	Te 52 (121.60)	I 53 (126.90)	Xe 54 (131.29)
Cs 55 (132.90)	Ba 56 (137.33)	La 57 (138.90)	Hf 72 (178.49)	Ta 73 (180.95)	W 74 (183.85)	Re 75 (186.21)	Os 76 (190.20)	Ir 77 (192.22)	Pt 78 (195.08)	Au 79 (196.97)	Hg 80 (200.59)	Tl 81 (204.38)	Pb 82 (207.20)	Bi 83 (208.98)	Po 84 (209.99)	At 85 (222.02)	Rn 86 (222.02)
Fr 87 (223.02)	Ra 88 (226.02)	Ac 89 (227.03)	Rf 104 (261.10)	Db 105 (262.11)	Sg 106 (263.12)	Bh 107 (264.12)	Hs 108 (277.13)	Mt 109 (288.14)	Ds 110 (288.14)	Rg 111 (288*)	Cn 112 (288*)		Fl 114 (289*)		Lv 116 (297*)		

• (4) F- ब्लॉक तत्व (F-block Elements)

Ce 58 (140.11)	Pr 59 (140.91)	Nd 60 (144.24)	Pm 61 (146.92)	Sm 62 (150.36)	Eu 63 (151.96)	Gd 64 (157.25)	Tb 65 (158.93)	Dy 66 (162.50)	Ho 67 (164.93)	Er 68 (167.26)	Tm 69 (168.93)	Yb 70 (173.04)	Lu 71 (174.97)				
Th 90 (232.04)	Pa 91 (231.04)	U 92 (238.05)	Np 93 (237.05)	Pu 94 (244.08)	Am 95 (243.08)	Cm 96 (247.07)	Bk 97 (247.07)	Cf 98 (251.08)	Es 99 (252.08)	Fm 100 (257.10)	Md 101 (259.10)	No 102 (259.11)	Lr 103 (262.11)				



## s-ब्लॉक तत्व (s-Block Elements)

s-उपकोण में अधिकतम दो इलेक्ट्रॉन रह सकते हैं।

→ A maximum of **two electrons** can remain in the s-subshell.

इसी कारण इस ब्लॉक में केवल **दो वर्ग (Group 1 और Group 2)** होते हैं।

→ Therefore, this block contains only **two groups (Group 1 and Group 2)**.

H	
1 (1.01)	
Li	Be
3 (6.94)	4 (9.01)
Na	Mg
11 (22.99)	12 (24.30)
K	Ca
19 (39.10)	20 (40.08)
Rb	Sr
37 (85.47)	38 (87.62)
Cs	Ba
55 (132.90)	56 (137.33)
Fr	Ra
87 (223.02)	88 (226.02)



## ■ ग्रुप 1 (Group 1)

इस समूह में **क्षारीय धातुएँ (Alkali Metals)** आती हैं।

→ **Alkali metals** belong to this group.

इनका इलेक्ट्रॉनिक विन्यास  $ns^1$  होता है।

→ Their electronic configuration is  $ns^1$ .

## ■ ग्रुप 2 (Group 2)

इस समूह में **क्षारीय मृदा धातुएँ (Alkaline Earth Metals)** आती हैं।

→ **Alkaline earth metals** belong to this group.

इनका इलेक्ट्रॉनिक विन्यास  $ns^2$  होता है।

→ Their electronic configuration is  $ns^2$ .

H	
1 (1.01)	
Li	Be
3 (6.94)	4 (9.01)
Na	Mg
11 (22.99)	12 (24.30)
K	Ca
19 (39.10)	20 (40.08)
Rb	Sr
37 (85.47)	38 (87.62)
Cs	Ba
55 (132.90)	56 (137.33)
Fr	Ra
87 (223.02)	88 (226.02)



## (1) s-ब्लॉक तत्त्व (s-Block Elements)



- Ha Li Na K Rb Csey Fryad

• Lithium (Li), Sodium (Na), Potassium (K), Rubidium (Ru), Caesium (Cs), and Francium (Fr).



### Beta Mange (Mg) Car Scoter (Sr) Bap Razi (Ra)

• Beryllium (Be), Magnesium (Mg), Calcium (Ca), Strontium (Sr), Barium (Br), and Radium (Ra).

S Block	
1	1.0078 H Hydrogen
2	3 6.9410 Li Lithium
3	11 22.990 Na Sodium
4	12 24.305 Mg Magnesium
5	19 39.098 K Potassium
6	20 40.078 Ca Calcium
7	37 85.468 Rb Rubidium
	38 87.620 Sr Strontium
6	55 132.91 Cs Caesium
	56 137.33 Ba Barium
7	87 223 Fr Francium
	88 226 Ra Radium



3  
**Li**  
Lithium  
6.94

11  
**Na**  
Sodium  
22.990

19  
**K**  
Potassium  
39.098

37  
**Rb**  
Rubidium  
85.468

55  
**Cs**  
Cesium  
132.905

87  
**Fr**  
Francium  
223.020





# ALKALI METALS STORED IN KEROSENE OIL



Sodium



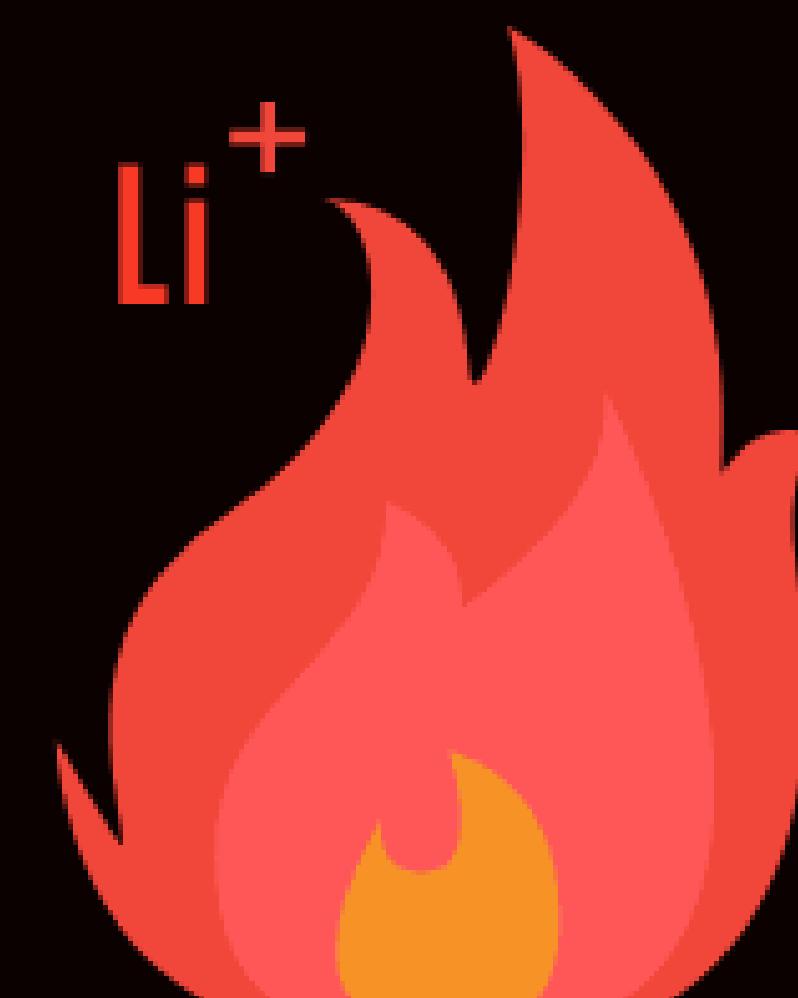
Lithium



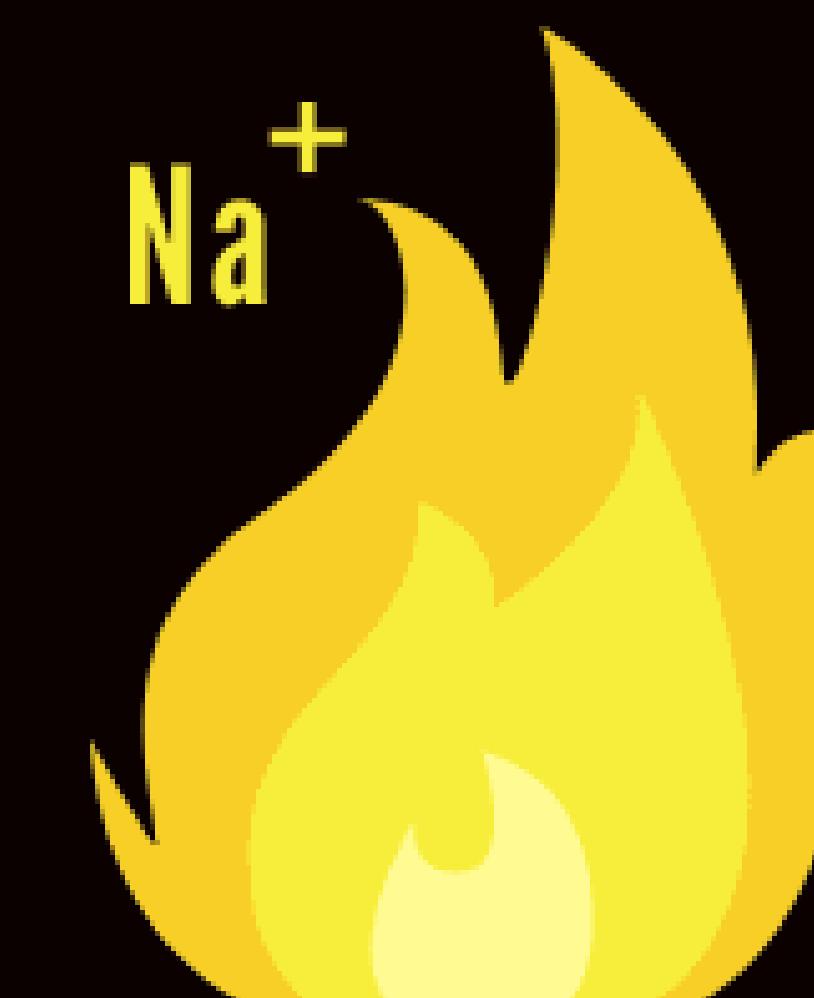
Potassium

# FLAME TEST

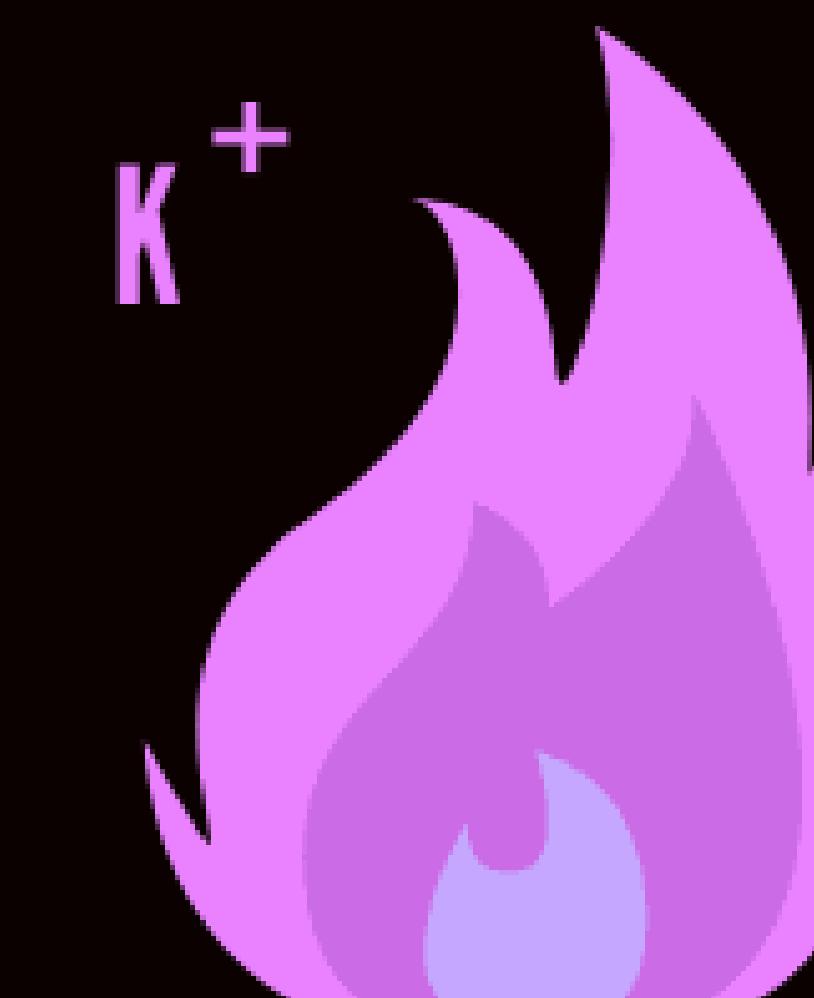
In chemistry, the flame test is an analytical procedure for identifying elements.



Lithium



Sodium



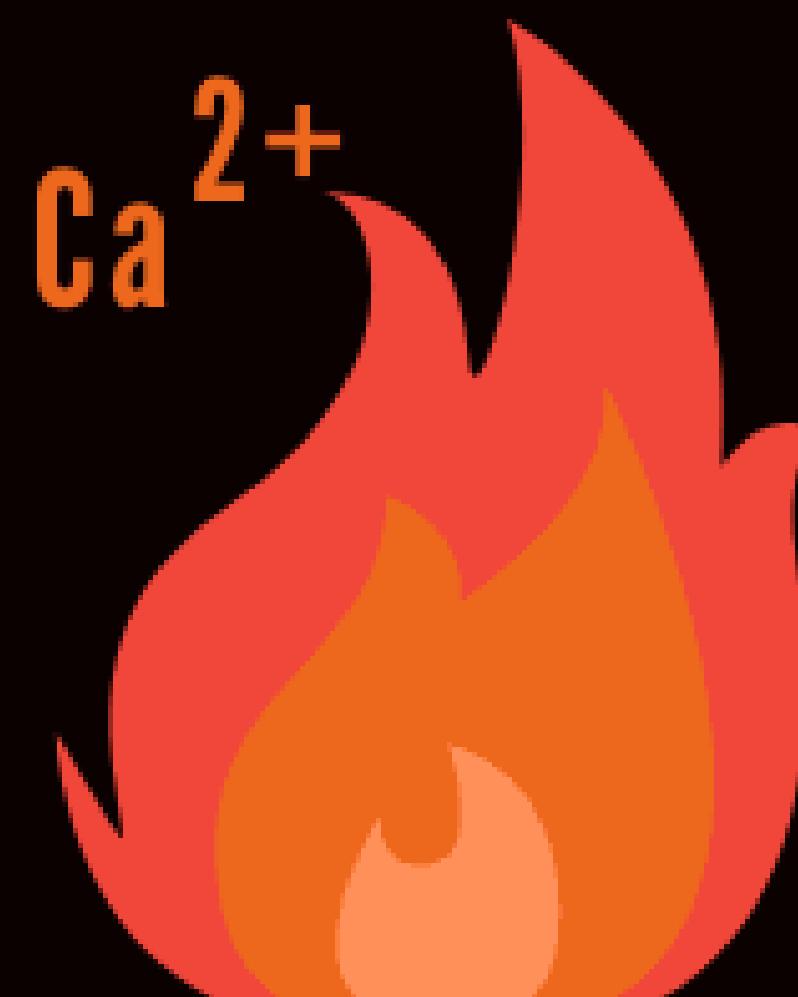
Potassium



Rubidium



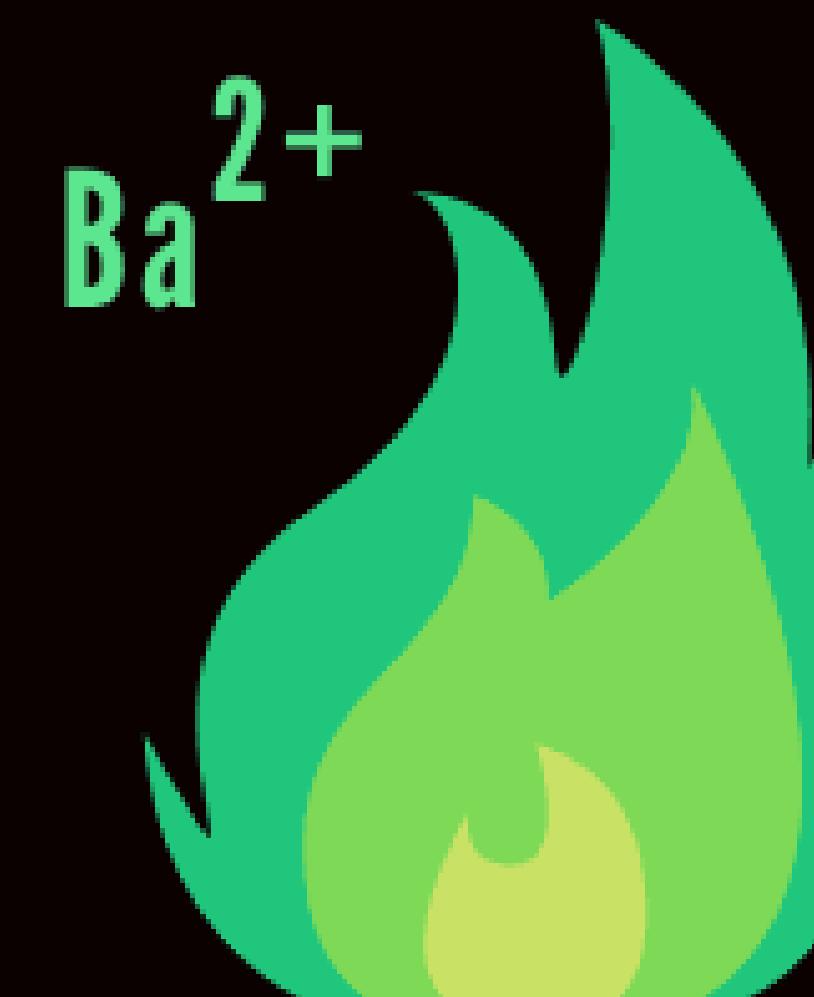
Cesium



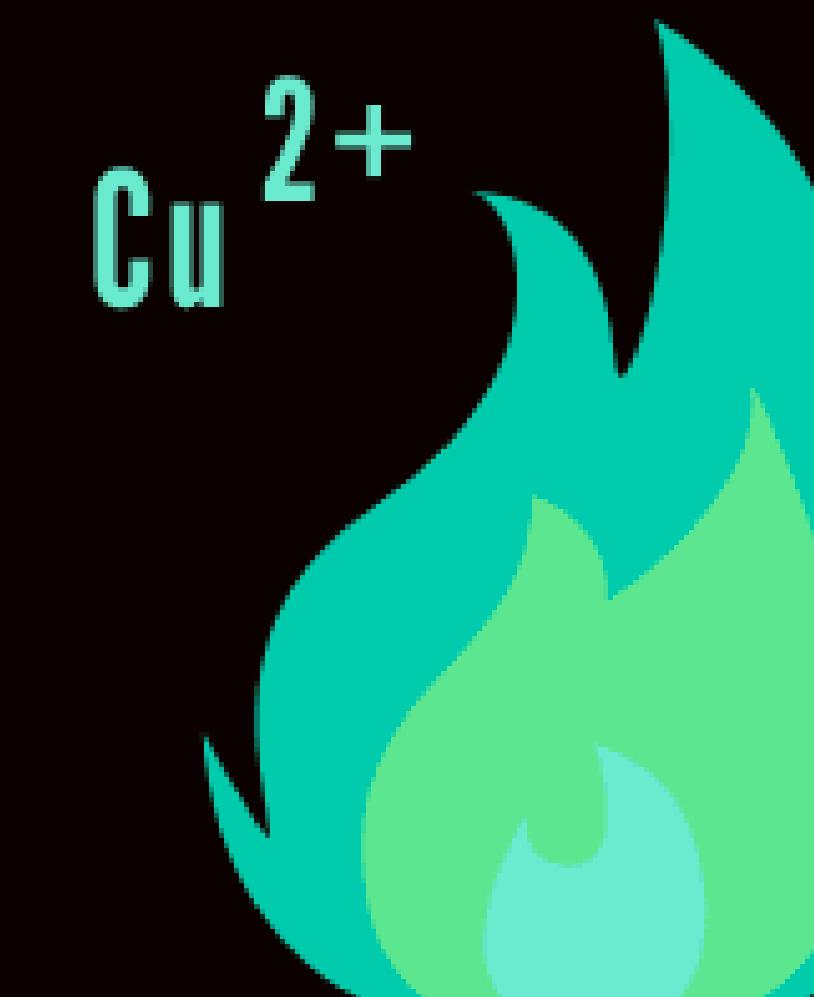
Calcium



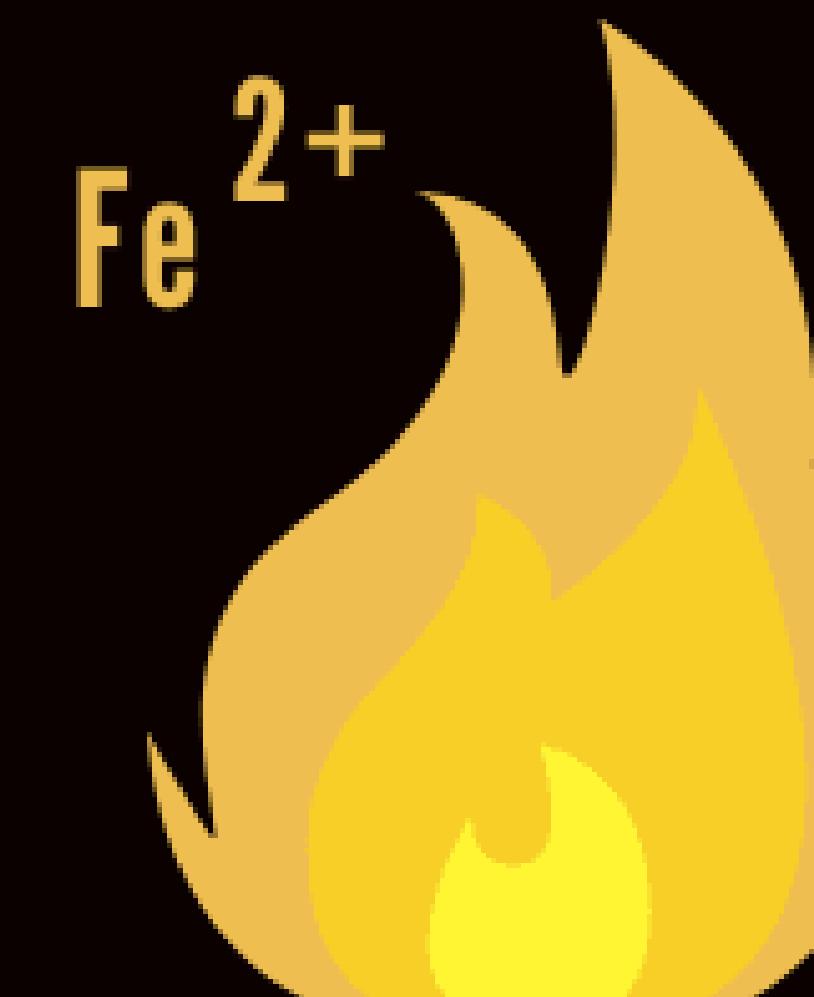
Strontium



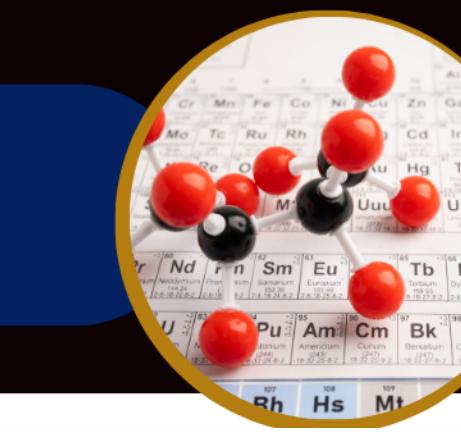
Barium



Copper



Iron



1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18			
<b>S Block</b>										<b>P Block</b>										
1 1.0078 <b>H</b> Hydrogen	2 6.9410 <b>Li</b> Lithium	3 9.0122 <b>Be</b> Beryllium	4 11.22990 <b>Na</b> Sodium	5 12.4305 <b>Mg</b> Magnesium	<b>D Block</b>										6 10.811 <b>B</b> Boron	7 12.011 <b>C</b> Carbon	8 14.007 <b>N</b> Nitrogen	9 15.999 <b>O</b> Oxygen	10 18.998 <b>F</b> Fluorine	2 20.180 <b>He</b> Helium
2 3 19.098 <b>K</b> Potassium	4 40.078 <b>Ca</b> Calcium	5 44.956 <b>Sc</b> Scandium	6 47.867 <b>Ti</b> Titanium	7 50.942 <b>V</b> Vanadium	8 51.996 <b>Cr</b> Chromium	9 54.938 <b>Mn</b> Manganese	10 55.845 <b>Fe</b> Iron	11 58.933 <b>Co</b> Cobalt	12 58.693 <b>Ni</b> Nickel	13 63.546 <b>Cu</b> Copper	14 65.380 <b>Zn</b> Zinc	15 69.723 <b>Al</b> Aluminium	16 72.640 <b>Si</b> Silicon	17 74.922 <b>P</b> Phosphorus	18 78.960 <b>S</b> Sulfur	19 79.904 <b>Cl</b> Chlorine	20 83.798 <b>Ar</b> Argon			
3 4 37.85.468 <b>Rb</b> Rubidium	5 87.620 <b>Sr</b> Strontium	6 88.906 <b>Y</b> Yttrium	7 91.224 <b>Zr</b> Zirconium	8 92.906 <b>Nb</b> Niobium	9 95.950 <b>Mo</b> Molybdenum	10 98 <b>Tc</b> Technetium	11 101.07 <b>Ru</b> Ruthenium	12 102.91 <b>Rh</b> Rhodium	13 106.42 <b>Pd</b> Palladium	14 107.87 <b>Ag</b> Silver	15 112.41 <b>Cd</b> Cadmium	16 114.82 <b>In</b> Indium	17 118.71 <b>Sn</b> Tin	18 121.76 <b>Sb</b> Antimony	19 127.60 <b>Te</b> Tellurium	20 126.90 <b>I</b> Iodine	21 131.29 <b>Xe</b> Xenon			
5 6 55.132.91 <b>Cs</b> Caesium	7 56.137.33 <b>Ba</b> Barium	8 57-71 57-71	9 72.178.49 <b>Hf</b> Hafnium	10 73.180.95 <b>Ta</b> Tantalum	11 74.183.84 <b>W</b> Tungsten	12 75.186.21 <b>Re</b> Rhenium	13 76.190.23 <b>Os</b> Osmium	14 77.192.22 <b>Ir</b> Iridium	15 78.195.08 <b>Pt</b> Platinum	16 79.196.97 <b>Au</b> Gold	17 80.200.59 <b>Hg</b> Mercury	18 81.204.38 <b>Tl</b> Thallium	19 82.207.20 <b>Pb</b> Lead	20 83.208.98 <b>Bi</b> Bismuth	21 84.209 <b>Po</b> Polonium	22 85.210 <b>At</b> Astatine	23 86.222 <b>Rn</b> Radon			
7 8 87.223 <b>Fr</b> Francium	9 88.226 <b>Ra</b> Radium	10 89-103 89-103	11 104.261 <b>Rf</b> Rutherfordium	12 105.262 <b>Db</b> Dubnium	13 106 <b>Sg</b> Seaborgium	14 107.264 <b>Bh</b> Bohrium	15 108.269 <b>Hs</b> Hassium	16 109.278 <b>Mt</b> Meitnerium	17 110.281 <b>Ds</b> Darmstadtium	18 111.282 <b>Rg</b> Roentgenium	19 112.285 <b>Cn</b> Copernicium	20 113.286 <b>Nh</b> Nihonium	21 114.289 <b>Fl</b> Flerovium	22 115.289 <b>Mc</b> Moscovium	23 116.293 <b>Lv</b> Livermorium	24 117.293 <b>Ts</b> Tennessine	25 118.294 <b>Og</b> Oganesson			

Atomic Number      1      1.0078      Atomic Mass  
 Element Symbol      H      Element Name      Hydrogen



## p-ब्लॉक तत्व (p-Block Elements)

ये वे तत्व होते हैं जिनमें अंतिम इलेक्ट्रॉन संयोजी **p-उपकोश्च (valence p-subshell)** में भरे जाते हैं।

13    14    15    16    17    18

👉 → These are the elements in which the last electron enters the **valence p-subshell**.

p-उपकोश्च अधिकतम **6 इलेक्ट्रॉन** रख सकता है।

👉 → The **p-subshell can hold a maximum of 6 electrons.**

इस कारण इस ब्लॉक में कुल **6 वर्ग (Groups)** होते हैं,

👉 जो कि **वर्ग 13 से वर्ग 18 तक** होते हैं।

→ Therefore, this block contains **6 groups**,

which are **Group 13 to Group 18.**

5 <b>B</b>	6 <b>C</b>	7 <b>N</b>	8 <b>O</b>	9 <b>F</b>	10 <b>Ne</b>
13 <b>Al</b>	14 <b>Si</b>	15 <b>P</b>	16 <b>S</b>	17 <b>Cl</b>	18 <b>Ar</b>
31 <b>Ga</b>	32 <b>Ge</b>	33 <b>As</b>	34 <b>Se</b>	35 <b>Br</b>	36 <b>Kr</b>
49 <b>In</b>	50 <b>Sn</b>	51 <b>Sb</b>	52 <b>Te</b>	53 <b>I</b>	54 <b>Xe</b>
81 <b>Tl</b>	82 <b>Pb</b>	83 <b>Bi</b>	84 <b>Po</b>	85 <b>At</b>	86 <b>Rn</b>
113 <b>Nh</b>	114 <b>Fl</b>	115 <b>Mc</b>	116 <b>Lv</b>	117 <b>Ts</b>	118 <b>Og</b>



## (2) p-ब्लॉक तत्त्व (p-Block Elements) -

13 14 15 16 17 18

Group 13 is known as **Boron Group**.

Boron (B), Aluminium (Al), Gallium (Ga), Indium (In),  
and Thallium (Tl).

**Began, Aaloo, Gajar In Thela**



5 <b>B</b>	6 <b>C</b>	7 <b>N</b>	8 <b>O</b>	9 <b>F</b>	10 <b>Ne</b>
13 <b>Al</b>	14 <b>Si</b>	15 <b>P</b>	16 <b>S</b>	17 <b>Cl</b>	18 <b>Ar</b>
31 <b>Ga</b>	32 <b>Ge</b>	33 <b>As</b>	34 <b>Se</b>	35 <b>Br</b>	36 <b>Kr</b>
49 <b>In</b>	50 <b>Sn</b>	51 <b>Sb</b>	52 <b>Te</b>	53 <b>I</b>	54 <b>Xe</b>
81 <b>Tl</b>	82 <b>Pb</b>	83 <b>Bi</b>	84 <b>Po</b>	85 <b>At</b>	86 <b>Rn</b>
113 <b>Nh</b>	114 <b>Fl</b>	115 <b>Mc</b>	116 <b>Lv</b>	117 <b>Ts</b>	118 <b>Og</b>



## (2) p-ब्लॉक तत्त्व (p-Block Elements) -

13    14    15    16    17    18

Group 14 is known as the **Carbon Group**.

Carbon (C), Silicon (Si), Germanium (Ge), Tin (Sn), and Lead (Pb).



- Chemistry Sir Gives Sanki Problems
- Chemistry Sir Gaye Sunday Pub Main
- Ca Si geye Tin (Sn) Lene

						2 <b>He</b>
5 <b>B</b>	6 <b>C</b>	7 <b>N</b>	8 <b>O</b>	9 <b>F</b>		10 <b>Ne</b>
13 <b>Al</b>	14 <b>Si</b>	15 <b>P</b>	16 <b>S</b>	17 <b>Cl</b>		18 <b>Ar</b>
31 <b>Ga</b>	32 <b>Ge</b>	33 <b>As</b>	34 <b>Se</b>	35 <b>Br</b>		36 <b>Kr</b>
49 <b>In</b>	50 <b>Sn</b>	51 <b>Sb</b>	52 <b>Te</b>	53 <b>I</b>		54 <b>Xe</b>
81 <b>Tl</b>	82 <b>Pb</b>	83 <b>Bi</b>	84 <b>Po</b>	85 <b>At</b>		86 <b>Rn</b>
113 <b>Nh</b>	114 <b>Fl</b>	115 <b>Mc</b>	116 <b>Lv</b>	117 <b>Ts</b>		118 <b>Og</b>



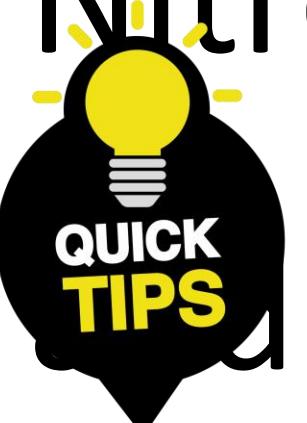
## (2) p-ब्लॉक तत्त्व (p-Block Elements) -

13 14 15 16 17 18

Group 15 is known as the **Nitrogen Group**. निकोजन्स

**(Pnicogens)**

Nitrogen (N), Phosphorus (P), Arsenic (As), Antimony (Sb),  
Bismuth (Bi).



- Nahi Pyaare Aise Sab Bigad Jaayega

- Nana Patekar Aishwariya SaB Bimar

5 <b>B</b>	6 <b>C</b>	7 <b>N</b>	8 <b>O</b>	9 <b>F</b>	10 <b>Ne</b>
13 <b>Al</b>	14 <b>Si</b>	15 <b>P</b>	16 <b>S</b>	17 <b>Cl</b>	18 <b>Ar</b>
31 <b>Ga</b>	32 <b>Ge</b>	33 <b>As</b>	34 <b>Se</b>	35 <b>Br</b>	36 <b>Kr</b>
49 <b>In</b>	50 <b>Sn</b>	51 <b>Sb</b>	52 <b>Te</b>	53 <b>I</b>	54 <b>Xe</b>
81 <b>Tl</b>	82 <b>Pb</b>	83 <b>Bi</b>	84 <b>Po</b>	85 <b>At</b>	86 <b>Rn</b>
113 <b>Nh</b>	114 <b>Fl</b>	115 <b>Mc</b>	116 <b>Lv</b>	117 <b>Ts</b>	118 <b>Og</b>



## (2) p-ब्लॉक तत्त्व (p-Block Elements) -

13 14 15 16 17 18

Group 16 is known as **Oxygen Group. कैल्कोजन (Chalcogen)**

Oxygen (O), Sulphur (S), Selenium (Se), Tellurium (Te), and the radioactive element Polonium (Po).



- Oh! Style Se Tel Polish Ker
- Oye Sun Saxena Teri Pol Kholu

5 B	6 C	7 N	8 O	9 F	10 Ne
13 Al	14 Si	15 P	16 S	17 Cl	18 Ar
31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr
49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe
81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn
113 Nh	114 Fl	115 Mc	116 Lv	117 Ts	118 Og



## (2) p-ब्लॉक तत्त्व (p-Block Elements) -

13 14 15 16 17 18

Group 17 is known as the group of **Halogens**.

Fluorine (F), Chlorine (Cl), Bromine (Br), Iodine (I), and Astatine (At).

QUICK  
TIPS

- Fufa Chachi Brother Inhone Atta Khaya
- First Class Biriyani In Austria

						2
						He
5	6	7	8	9		10
B	C	N	O	F		Ne
13	14	15	16	17		18
Al	Si	P	S	Cl		Ar
31	32	33	34	35		36
Ga	Ge	As	Se	Br		Kr
49	50	51	52	53		54
In	Sn	Sb	Te	I		Xe
81	82	83	84	85		86
Tl	Pb	Bi	Po	At		Rn
113	114	115	116	117		118
Nh	Fl	Mc	Lv	Ts		Og



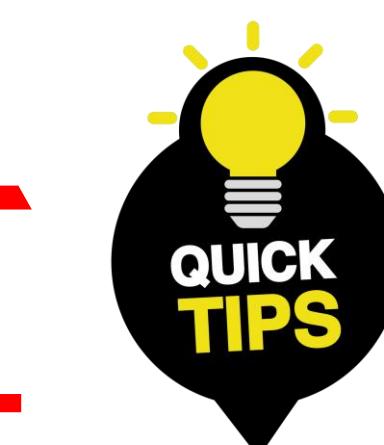
## (2) p-ब्लॉक तत्त्व (p-Block Elements) -

13 14 15 16 17 18

Group 18 is known as the group of **Noble Gases**,

Helium (He), Neon (Ne), Argon (Ar), Krypton (Kr), Xenon (Xe), and the radioactive Radon (Rn).

**हीना नीना और करीना एकसरे रंगीन**



He हीना

Ne नीना

Ar और

Kr करीना

Xe एकसरे

Rn रंगीन

5 B	6 C	7 N	8 O	9 F	10 Ne
13 Al	14 Si	15 P	16 S	17 Cl	18 Ar
31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr
49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe
81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn
113 Nh	114 Fl	115 Mc	116 Lv	117 Ts	118 Og



1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18			
<b>S Block</b>										<b>P Block</b>										
1 1.0078 <b>H</b> Hydrogen	2 6.9410 <b>Li</b> Lithium	3 9.0122 <b>Be</b> Beryllium	4 11.22990 <b>Na</b> Sodium	5 12.4305 <b>Mg</b> Magnesium	<b>D Block</b>										6 10.811 <b>B</b> Boron	7 12.011 <b>C</b> Carbon	8 14.007 <b>N</b> Nitrogen	9 15.999 <b>O</b> Oxygen	10 18.998 <b>F</b> Fluorine	11 20.180 <b>Ne</b> Neon
2 3 19.098 <b>K</b> Potassium	4 20.4078 <b>Ca</b> Calcium	5 21.44.956 <b>Sc</b> Scandium	6 22.47.867 <b>Ti</b> Titanium	7 23.50.942 <b>V</b> Vanadium	8 24.51.996 <b>Cr</b> Chromium	9 25.54.938 <b>Mn</b> Manganese	10 26.55.845 <b>Fe</b> Iron	11 27.58.933 <b>Co</b> Cobalt	12 28.58.693 <b>Ni</b> Nickel	13 29.63.546 <b>Cu</b> Copper	14 30.65.380 <b>Zn</b> Zinc	15 31.69.723 <b>Al</b> Aluminium	16 32.72.640 <b>Si</b> Silicon	17 33.74.922 <b>P</b> Phosphorus	18 34.78.960 <b>S</b> Sulfur	19 35.79.904 <b>Cl</b> Chlorine	20 36.83.798 <b>Ar</b> Argon			
3 4 37.85.468 <b>Rb</b> Rubidium	5 38.87.620 <b>Sr</b> Strontium	6 39.88.906 <b>Y</b> Yttrium	7 40.91.224 <b>Zr</b> Zirconium	8 41.92.906 <b>Nb</b> Niobium	9 42.95.950 <b>Mo</b> Molybdenum	10 43.98 <b>Tc</b> Technetium	11 44.101.07 <b>Ru</b> Ruthenium	12 45.102.91 <b>Rh</b> Rhodium	13 46.106.42 <b>Pd</b> Palladium	14 47.107.87 <b>Ag</b> Silver	15 48.112.41 <b>Cd</b> Cadmium	16 49.114.82 <b>In</b> Indium	17 50.118.71 <b>Sn</b> Tin	18 51.121.76 <b>Sb</b> Antimony	19 52.127.60 <b>Te</b> Tellurium	20 53.126.90 <b>I</b> Iodine	21 54.131.29 <b>Xe</b> Xenon			
5 6 55.132.91 <b>Cs</b> Caesium	7 56.137.33 <b>Ba</b> Barium	8 57-71 57-71	9 72.178.49 <b>Hf</b> Hafnium	10 73.180.95 <b>Ta</b> Tantalum	11 74.183.84 <b>W</b> Tungsten	12 75.186.21 <b>Re</b> Rhenium	13 76.190.23 <b>Os</b> Osmium	14 77.192.22 <b>Ir</b> Iridium	15 78.195.08 <b>Pt</b> Platinum	16 79.196.97 <b>Au</b> Gold	17 80.200.59 <b>Hg</b> Mercury	18 81.204.38 <b>Tl</b> Thallium	19 82.207.20 <b>Pb</b> Lead	20 83.208.98 <b>Bi</b> Bismuth	21 84.209 <b>Po</b> Polonium	22 85.210 <b>At</b> Astatine	23 86.222 <b>Rn</b> Radon			
7 8 87.223 <b>Fr</b> Francium	9 88.226 <b>Ra</b> Radium	10 89-103 89-103	11 104.261 <b>Rf</b> Rutherfordium	12 105.262 <b>Db</b> Dubnium	13 106 <b>Sg</b> Seaborgium	14 107.264 <b>Bh</b> Bohrium	15 108.269 <b>Hs</b> Hassium	16 109.278 <b>Mt</b> Meitnerium	17 110.281 <b>Ds</b> Darmstadtium	18 111.282 <b>Rg</b> Roentgenium	19 112.285 <b>Cn</b> Copernicium	20 113.286 <b>Nh</b> Nihonium	21 114.289 <b>Fl</b> Flerovium	22 115.289 <b>Mc</b> Moscovium	23 116.293 <b>Lv</b> Livermorium	24 117 <b>Ts</b> Tennessine	25 118.294 <b>Og</b> Oganesson			

Atomic Number  
Element Symbol  
Element Name

1      1.0078

**H**

Hydrogen



## D-ब्लॉक तत्व (D-BLOCK ELEMENTS).

👉 d-उपकोण अधिकतम **10 इलेक्ट्रॉन** रख सकता है।

→ The **d-subshell can hold a maximum of 10 electrons.**

👉 इस कारण इस ब्लॉक में कुल **10 वर्ग (Groups)** होते हैं,

जो कि **वर्ग 3 से वर्ग 12 तक** होते हैं।

→ Therefore, this block contains **10 groups**,

which are **Group 3 to Group 12.**

👉 इन्हें **संक्रमण तत्व (Transition Elements)** कहा जाता है।

→ These elements are called **Transition Elements.**

👉 सभी d-ब्लॉक तत्व **धातु (Metals)** होते हैं।

→ All d-block elements are **metals.**

1	2	3	4	5	6	7	8	9	10	11	12	13
S Block												
1	2	3	4	5	6	7	8	9	10	11	12	13
1 H Hydrogen	2 Li Lithium	3 Be Beryllium	4 Na Sodium	5 Mg Magnesium	6	7	8	9	10	11	12	13
19 K Potassium	20 Ca Calcium	21 Sc Scandium	22 Ti Titanium	23 V Vanadium	24 Cr Chromium	25 Mn Manganese	26 Fe Iron	27 Co Cobalt	28 Ni Nickel	29 Cu Copper	30 Zn Zinc	31 Ga Gallium
37 Rb Rubidium	38 Sr Strontium	39 Y Yttrium	40 Zr Zirconium	41 Nb Niobium	42 Mo Molybdenum	43 Tc Technetium	44 Ru Ruthenium	45 Rh Rhodium	46 Pd Palladium	47 Ag Silver	48 Cd Cadmium	49 In Indium
55 Cs Caesium	56 Ba Barium	57-71	72 Hf Hafnium	73 Ta Tantalum	74 W Tungsten	75 Re Rhenium	76 Os Osmium	77 Ir Iridium	78 Pt Platinum	79 Au Gold	80 Hg Mercury	81 Tl Thallium
87 Fr Francium	88 Ra Radium	89-103	104 Rf Rutherfordium	105 Db Dubnium	106 Sg Seaborgium	107 Bh Bohrium	108 Hs Hassium	109 Mt Meitnerium	110 Ds Darmstadtium	111 Rg Roentgenium	112 Cn Copernicium	113 Nh Nihonium



	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
S Block										P Block								
1	1 1.0078 <b>H</b> Hydrogen	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
2	3 6.9410 <b>Li</b> Lithium	4 9.0122 <b>Be</b> Beryllium																
3	11 22.990 <b>Na</b> Sodium	12 24.305 <b>Mg</b> Magnesium																
4	19 39.098 <b>K</b> Potassium	20 40.078 <b>Ca</b> Calcium	21 44.956 <b>Sc</b> Scandium	22 47.867 <b>Ti</b> Titanium	23 50.942 <b>V</b> Vanadium	24 51.996 <b>Cr</b> Chromium	25 54.938 <b>Mn</b> Manganese	26 55.845 <b>Fe</b> Iron	27 58.933 <b>Co</b> Cobalt	28 58.693 <b>Ni</b> Nickel	29 63.546 <b>Cu</b> Copper	30 65.380 <b>Zn</b> Zinc	31 69.723 <b>Ga</b> Gallium	32 72.640 <b>Ge</b> Germanium	33 74.922 <b>As</b> Arsenic	34 78.960 <b>Se</b> Selenium	35 79.904 <b>Br</b> Bromine	36 83.798 <b>Kr</b> Krypton
5	37 85.468 <b>Rb</b> Rubidium	38 87.620 <b>Sr</b> Strontium	39 88.906 <b>Y</b> Yttrium	40 91.224 <b>Zr</b> Zirconium	41 92.906 <b>Nb</b> Niobium	42 95.950 <b>Mo</b> Molybdenum	43 98 <b>Tc</b> Technetium	44 101.07 <b>Ru</b> Ruthenium	45 102.91 <b>Rh</b> Rhodium	46 106.42 <b>Pd</b> Palladium	47 107.87 <b>Ag</b> Silver	48 112.41 <b>Cd</b> Cadmium	49 114.82 <b>In</b> Indium	50 118.71 <b>Sn</b> Tin	51 121.76 <b>Sb</b> Antimony	52 127.60 <b>Te</b> Tellurium	53 126.90 <b>I</b> Iodine	54 131.29 <b>Xe</b> Xenon
6	55 132.91 <b>Cs</b> Caesium	56 137.33 <b>Ba</b> Barium	57 - 71	72 178.49 <b>Hf</b> Hafnium	73 180.95 <b>Ta</b> Tantalum	74 183.84 <b>W</b> Tungsten	75 186.21 <b>Re</b> Rhenium	76 190.23 <b>Os</b> Osmium	77 192.22 <b>Ir</b> Iridium	78 195.08 <b>Pt</b> Platinum	79 196.97 <b>Au</b> Gold	80 200.59 <b>Hg</b> Mercury	81 204.38 <b>Tl</b> Thallium	82 207.20 <b>Pb</b> Lead	83 208.98 <b>Bi</b> Bismuth	84 209 <b>Po</b> Polonium	85 210 <b>At</b> Astatine	86 222 <b>Rn</b> Radon
7	87 223 <b>Fr</b> Francium	88 226 <b>Ra</b> Radium	89 - 103	104 261 <b>Rf</b> Rutherfordium	105 262 <b>Db</b> Dubnium	106 <b>Sg</b> Seaborgium	107 264 <b>Bh</b> Bohrium	108 269 <b>Hs</b> Hassium	109 278 <b>Mt</b> Meitnerium	110 281 <b>Ds</b> Darmstadtium	111 282 <b>Rg</b> Roentgenium	112 285 <b>Cn</b> Copernicium	113 286 <b>Nh</b> Nihonium	114 289 <b>Fl</b> Flerovium	115 289 <b>Mc</b> Moscovium	116 293 <b>Lv</b> Livermorium	117 <b>Ts</b> Tennessine	118 294 <b>Og</b> Oganesson

Atomic Number  
Element Symbol  
Element Name

1 1.0078  
**H**  
Hydrogen

1 1.0078  
**H**

Hydrogen



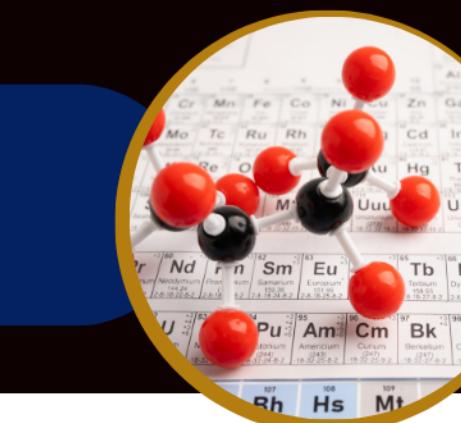
## Period 4 (चतुर्थ अवर्त)

यह अवधि मुख्य रूप से उन तत्वों से बनी होती है जो पृथकी की पपड़ी और कोर (crust and core) में पाए जाते हैं तथा अपनी स्थिरता (stability) के लिए प्रसिद्ध हैं।

**Scandium (Sc), Titanium (Ti), Vanadium (V), Chromium (Cr), Manganese (Mn), Iron (Fe), Cobalt (Co), Nickel (Ni), Copper (Cu), and Zinc (Zn).**

(Trick) ➡ "साइंस टीचर विवेक सर मांगे फेयर कॉपी नहीं तो क्लास से जाओ"

		1	2
1	H		He
2	Li	Be	
3	Na	Mg	
4	K	Ca	
5	Rb	Sr	
6	Sc	Ti	
7	V	Cr	
8	Mn	Fe	
9	Co	Ni	
10	Cu	Zn	
11	Ga	Ge	
12	As	As	
13	Si	In	
14	P	Sn	
15	S	Sb	
16	Cl	Te	
17	Ar	I	
18		Xe	
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# Period 5 (पंचम आवर्त)

**Yttrium (Y), Zirconium (Zr), Niobium (Nb), Molybdenum (Mo), Technetium (Tc), Ruthenium (Ru), Rhodium (Rh), Palladium (Pd), silver (Ag), और Cadmium (Cd).**

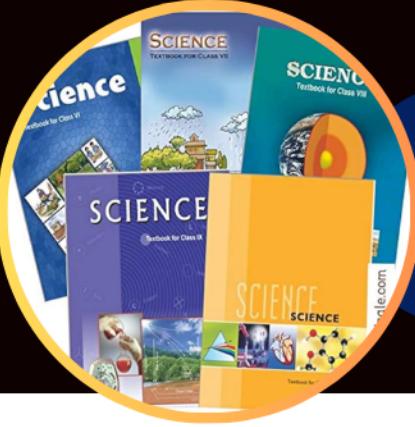
ये जिंदगी नहीं, मोहब्बत तेरी रोरोकर पुकारेगी आज चांदनी



**Lutetium (Lu), Hafnium (Hf), Tantalum (Ta), Tungsten (W), Rhenium (Re), Osmium (Os), Iridium (Ir),  
Platinum (Pt), Gold (Au), and Mercury (Hg).**

La हफ्ता करना रे उस आइरिस से पिटेगा और हाँस्पिटल जाएगा।

# La Hf Ta W Re Os Ir Pt Au Hg



<sup>1</sup> <b>H</b> Hydrogen															<sup>2</sup> <b>He</b> Helium		
<sup>3</sup> <b>Li</b> Lithium	<sup>4</sup> <b>Be</b> Boron																
<sup>11</sup> <b>Na</b> Sodium	<sup>12</sup> <b>Mg</b> Magnesium																
<sup>19</sup> <b>K</b> Potassium	<sup>20</sup> <b>Ca</b> Calcium	<sup>21</sup> <b>Sc</b> Scandium	<sup>22</sup> <b>Ti</b> Titanium	<sup>23</sup> <b>V</b> Vanadium	<sup>24</sup> <b>Cr</b> Chromium	<sup>25</sup> <b>Mn</b> Manganese	<sup>26</sup> <b>Fe</b> Iron	<sup>27</sup> <b>Co</b> Cobalt	<sup>28</sup> <b>Ni</b> Nickel	<sup>29</sup> <b>Cu</b> Copper	<sup>30</sup> <b>Zn</b> Zinc	<sup>31</sup> <b>Ga</b> Gallium	<sup>32</sup> <b>Ge</b> Germanium	<sup>33</sup> <b>As</b> Arsenic	<sup>34</sup> <b>Se</b> Selenium	<sup>35</sup> <b>Br</b> Bromine	<sup>36</sup> <b>Kr</b> Krypton
<sup>37</sup> <b>Rb</b> Rubidium	<sup>38</sup> <b>Sr</b> Strontium	<sup>39</sup> <b>Y</b> Yttrium	<sup>40</sup> <b>Zr</b> Zirconium	<sup>41</sup> <b>Nb</b> Niobium	<sup>42</sup> <b>Mo</b> Molybdenum	<sup>43</sup> <b>Tc</b> Technetium	<sup>44</sup> <b>Ru</b> Ruthenium	<sup>45</sup> <b>Rh</b> Rhodium	<sup>46</sup> <b>Pd</b> Palladium	<sup>47</sup> <b>Ag</b> Silver	<sup>48</sup> <b>Cd</b> Cadmium	<sup>49</sup> <b>In</b> Indium	<sup>50</sup> <b>Sn</b> Tin	<sup>51</sup> <b>Sb</b> Antimony	<sup>52</sup> <b>Te</b> Tellurium	<sup>53</sup> <b>I</b> Iodine	<sup>54</sup> <b>Xe</b> Xenon
<sup>55</sup> <b>Cs</b> Cesium	<sup>56</sup> <b>Ba</b> Barium	<sup>57</sup> <b>La</b> Lanthanum	<sup>72</sup> <b>Hf</b> Hafnium	<sup>73</sup> <b>Ta</b> Tantalum	<sup>74</sup> <b>W</b> Tungsten	<sup>75</sup> <b>Re</b> Rhenium	<sup>76</sup> <b>Os</b> Osmium	<sup>77</sup> <b>Ir</b> Iridium	<sup>78</sup> <b>Pt</b> Platinum	<sup>79</sup> <b>Au</b> Gold	<sup>80</sup> <b>Hg</b> Mercury	<sup>81</sup> <b>Tl</b> Thallium	<sup>82</sup> <b>Pb</b> Lead	<sup>83</sup> <b>Bi</b> Bismuth	<sup>84</sup> <b>Po</b> Polonium	<sup>85</sup> <b>At</b> Astatine	<sup>86</sup> <b>Rn</b> Radon
<sup>87</sup> <b>Fr</b> Francium	<sup>88</sup> <b>Ra</b> Radium	<sup>89</sup> <b>Ac</b> Actinium	<sup>104</sup> <b>Rf</b> Rutherfordium	<sup>105</sup> <b>Db</b> Dubnium	<sup>106</sup> <b>Sg</b> Seaborgium	<sup>107</sup> <b>Bh</b> Bohrium	<sup>108</sup> <b>Hs</b> Hassium	<sup>109</sup> <b>Mt</b> Meitnerium	<sup>110</sup> <b>Ds</b> Darmstadtium	<sup>111</sup> <b>Rg</b> Roentgenium	<sup>112</sup> <b>Cn</b> Copernicium	<sup>113</sup> <b>Nh</b> Nihonium	<sup>114</sup> <b>F</b> Florium	<sup>115</sup> <b>Mc</b> Moscovium	<sup>116</sup> <b>Lv</b> Livermorium	<sup>117</sup> <b>Ts</b> Tennessee	<sup>118</sup> <b>Og</b> Oganesson

<sup>58</sup> <b>Ce</b> Cerium	<sup>59</sup> <b>Pr</b> Praseodymium	<sup>60</sup> <b>Nd</b> Neodymium	<sup>61</sup> <b>Pm</b> Promethium	<sup>62</sup> <b>Sm</b> Samarium	<sup>63</sup> <b>Eu</b> Europium	<sup>64</sup> <b>Gd</b> Gadolinium	<sup>65</sup> <b>Tb</b> Terbium	<sup>66</sup> <b>Dy</b> Dysprosium	<sup>67</sup> <b>Ho</b> Holmium	<sup>68</sup> <b>Er</b> Erbium	<sup>69</sup> <b>Tm</b> Thulium	<sup>70</sup> <b>Yb</b> Ytterbium	<sup>71</sup> <b>Lu</b> Lutetium
<sup>90</sup> <b>Th</b> Thorium	<sup>91</sup> <b>Pa</b> Protactinium	<sup>92</sup> <b>U</b> Uranium	<sup>93</sup> <b>Np</b> Neptunium	<sup>94</sup> <b>Pu</b> Plutonium	<sup>95</sup> <b>Am</b> Americium	<sup>96</sup> <b>Cm</b> Curium	<sup>97</sup> <b>Bk</b> Berkelium	<sup>98</sup> <b>Cf</b> Californium	<sup>99</sup> <b>Es</b> Einsteinium	<sup>100</sup> <b>Fm</b> Fermium	<sup>101</sup> <b>Md</b> Mendelevium	<sup>102</sup> <b>No</b> Nobelium	<sup>103</sup> <b>Lr</b> Lawrencium



# आवर्त सारणी (Periodic Table) में -

# 6th Period

 इसे लैंथेनाइड श्रृंखला (Lanthanide Series) या Rare Earth Series कहा जाता है।

# क्योंकि:

इसमें तत्व Lanthanum (La) से लेकर Lutetium (Lu) तक होते हैं (Atomic number 57-71)।

इन तत्वों को “दुर्लभ पृथकी तत्व (Rare Earth Elements)” भी कहा जाता है

# 7th Period

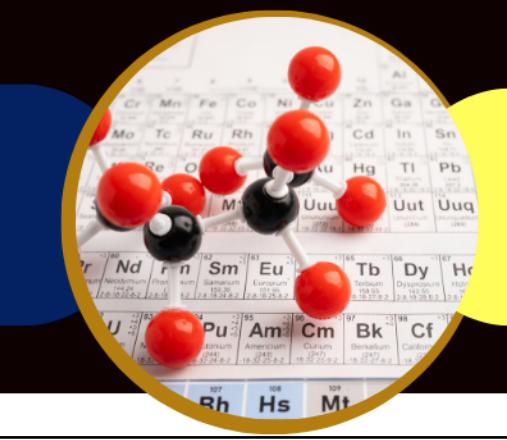
## इसे ऐक्टिनाइड श्रृंखला (Actinide Series) कहा जाता है

इसमें तत्व **Actinium (Ac)** से लेकर **Lawrencium (Lr)** तक होते हैं (Atomic number 89-103)।

इन तत्वों को “**Radioactive Elements (रेडियोधर्मी तत्व)**” भी कहा जाता है, क्योंकि अधिकांश तत्व रेडियोधर्मी होते हैं।

# Periodic Table of the Elements

	57	La	Cerium 140.116	58	Ce	Praseodymium 140.908	59	Pr	Neodymium 144.243	60	Nd	Promethium	144.913	61	Pm	Holmium 167.259	62	Sm	Dysprosium 158.925	63	Eu	Terbium 162.500	64	Gd	Thulium 151.964	65	Tb	Europium 157.25	66	Ho	Gadolinium 168.934	67	Er	Ytterbium 150.36	68	Dy	Lutetium 173.055	69	Tm	Erbium 174.967	70	Yb		71	Lu	
	89	Ac	Actinium 227.028	90	Th	Thorium 232.038	91	Pa	Protactinium 231.036	92	U	Uranium 238.029	93	Np	Neptunium 237.048	94	Cf	Plutonium 243.061	95	Bk	Americium 247.070	96	Fm	Berkelium 251.080	97	Es	Einsteinium 257.095	98	Md	Mendelevium [254]	99	Pu	Curium 247.070	100	Cm	Californium 259.101	101	No	Nobelium 258.1	102	Lr	Lawrencium [262]	103			



तथ्य / Fact	तत्व (Element)
□ कुल ज्ञात तत्व / Total Known Elements	118
□ प्रकृति में प्राप्त तत्व / Naturally Occurring Elements	94
□ कृत्रिम विधि से निर्मित तत्व / Artificially Prepared Elements	24
□ धातु तत्वों की संख्या / Number of Metallic Elements	91
□ अधातु तत्वों की संख्या / Number of Non-Metallic Elements	20
□ उपधातु तत्वों की संख्या / Number of Metalloids	7
□ भूपर्फटी (Earth Crust) में सर्वाधिक मात्रा में पाया जाने वाला तत्व	ऑक्सीजन (Oxygen) – 8
□ Most Abundant Element in Earth Crust	
□ भूपर्फटी में सर्वाधिक मात्रा में पाया जाने वाला धातु तत्व	एल्युमिनियम (Aluminium) – 13
□ Most Abundant Metal in Earth Crust	
□ सबसे हल्का तत्व / Lightest Element	हाइड्रोजन (Hydrogen) – 1
□ सबसे भारी तत्व / Heaviest Element	ऑस्मियम (Osmium) – 76
□ सबसे हल्का धातु तत्व / Lightest Metal	लिथियम (Lithium) – 3

# Periodic Table of the Elements



□ द्रव धातु तत्व / Liquid Metal Element	पारा (Mercury) – <b>80</b>
□ द्रव अधातु तत्व / Liquid Non-Metal Element	ब्रोमीन (Bromine) – <b>35</b>
□ विद्युत की सबसे अच्छी सुचालक धातु तत्व / Best Electrical Conductor Metal	चाँदी (Silver) – <b>47</b>
□ विद्युत का सुचालक अधातु तत्व / Conducting Non-Metal	ग्रेफाइट (Graphite - कार्बन Carbon) – <b>6</b>
□ सबसे अधिक आघातवर्धनीय (Malleable) तत्व / Most Malleable Metal	सोना (Gold) – <b>79</b>
□ सबसे अधिक क्रियाशील अधातु तत्व / Most Reactive Non-Metal	फ्लोरीन (Fluorine) – <b>9</b>
□ सबसे अधिक क्रियाशील धातु तत्व / Most Reactive Metal	सीज़ियम (Cesium) – <b>55</b>
□ सर्वाधिक आयनीकरण विभव वाला तत्व / Highest Ionization Potential Element	हीलियम (Helium) – <b>2</b>
□ न्यूनतम आयनीकरण विभव वाला तत्व / Lowest Ionization Potential Element	सीज़ियम (Cesium) – <b>55</b>
□ सर्वाधिक इलेक्ट्रॉन ग्रहण करने वाला तत्व / Highest Electron Affinity Element	क्लोरीन (Chlorine) – <b>17</b>
□ सर्वाधिक विद्युत क्रृणात्मक तत्व / Most Electronegative Element	फ्लोरीन (Fluorine) – <b>9</b>
□ सबसे प्रबल ऑक्सीकारक तत्व / Strongest Oxidizing Element	फ्लोरीन (Fluorine) – <b>9</b>
□ मानव शरीर में सर्वाधिक मात्रा में पाया जाने वाला तत्व /	ऑक्सीजन (Oxygen) – <b>8</b>
□ Most Abundant Element in Human Body	